

TANDEM MICHAEL REACTIONS FOR THE CONSTRUCTION OF PYRROLIDINE AND PIPERIDINE RING SYSTEMS

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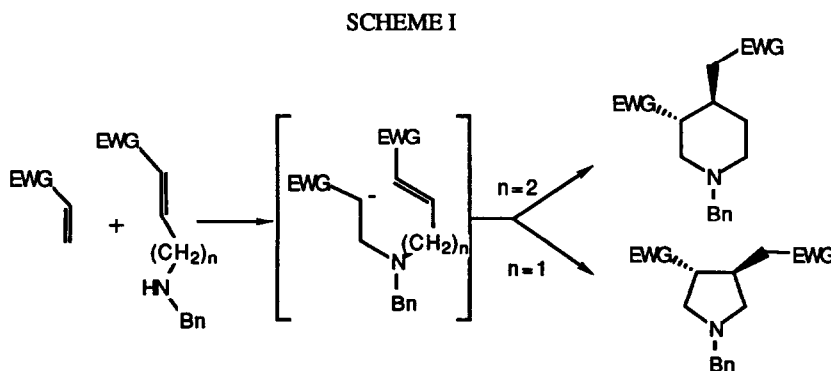
Summary: A convergent one-pot construction of disubstituted pyrrolidine and piperidine frameworks has been accomplished through a simple protocol involving intermolecular conjugate addition of a nitrogen nucleophile to an electrophilic olefin followed by intramolecular trapping of the generated enolate by a built-in α,β -unsaturated acceptor.

Many natural products contain pyrrolidine and piperidine substructures and so the development of methods for preparing five- and six-membered nitrogen heterocycles has received much attention.

The most common methodologies utilized to gain access to these materials entail pericyclic reactions such as the ene reaction¹, 1,3-dipolar cycloadditions² and sigmatropic rearrangements³ as well as intramolecular cationic⁴, anionic⁵ or radical⁶ promoted heterocyclizations.

Recently, several groups have reported efficient variants of Michael-Michael ring closure procedures for the preparation of functionalized cyclic rings⁷. However, most of this work is confined to the cyclization by carbon nucleophiles, less attention being directed towards nitrogen nucleophiles.

As a part of our ongoing interest in the area of neuroexcitatory amino acids we were intrigued to evaluate the feasibility of a conceptually simple approach to five- and six-membered nitrogen heterocycles involving intermolecular conjugate addition of a nitrogen nucleophile to an acceptor molecule to generate an enolate intermediate, which is subsequently captured intramolecularly by the built-in α,β -unsaturated acceptor leading to cyclization as indicated in the Scheme I.

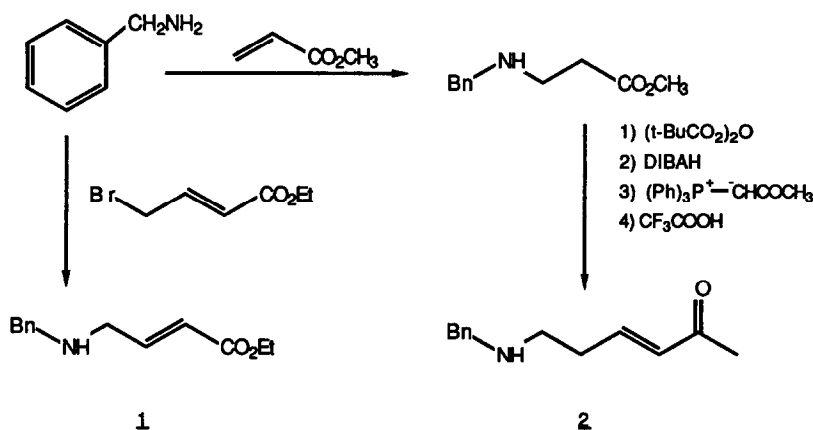


In this letter we report the successful execution of this strategy which allows a convergent construction of substituted pyrrolidine and piperidine frameworks in a very convenient one-pot procedure.

According to the Scheme I, the opening step called for the preparation of two subunits, such as **1** and **2**, containing both the Michael donor nitrogen nucleophile and a suitably located acceptor moiety.

This task has been accomplished in a straightforward manner for **1** through the reaction of benzylamine with ethyl 4-bromo-2-butenate, while the preparation of **2** has been realized, again starting from benzylamine, by standard steps as illustrated in the Scheme II.

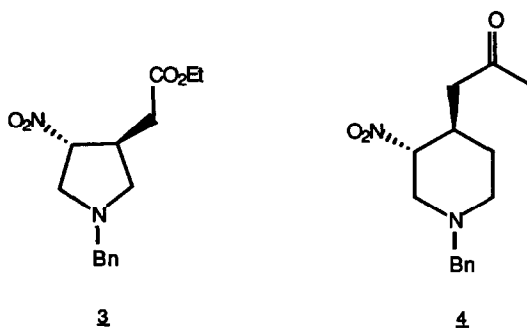
SCHEME II



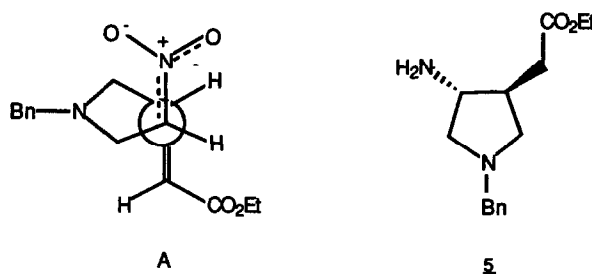
We chose secondary benzylated amines as partners for cyclization since preliminary experiments indicated that primary amines afford mixture of mono- and bis-adducts in the intermolecular step. Moreover, chiral building-blocks could be derived by alkylation of readily available optically active phenethylamines.

The second donor-acceptor subunit should be characterized by a carbon-carbon double bond conjugated with an electron-withdrawing group, which could be transformed or removed at a later stage after heterocyclic ring formation.

The well known versatility of the nitro group suggested the choice of nitroethylene for a model study, its propensity to anionic polymerization in the presence of aliphatic amines being overcome by its generation *in situ* from 1-benzoyloxy-2-nitroethane⁸. Thus, treatment of equimolecular amounts of the latter with both **1** and **2** at room temperature led to the direct formation of the corresponding cyclization products **3** and **4** in good yield after flash chromatography⁹.



The stereospecific course¹⁰ of the sequence leading to a trans-arrangement of the two substituents in the heterocycle can be explained by the intramolecular Michael addition taking place through the transition state A, depicted for the formation of **3**, with an antiperiplanar orientation between the nitro group and the acceptor chain¹¹. The same stereochemical outcome has been observed by Bunce et al^{7b} in related carbocyclizations.



Moreover, chemical proof of the trans-relationship of the nitro group and the acetic chain rests on the reluctance of the derived amino-ester **5** to undergo intramolecular lactamization, in sharp contrast with the facile aminolysis of γ -aminoesters to form γ -butyrolactams.

In summary, the present methodology provides a new and promising access to functionalized pyrrolidines and piperidines, which could be extended to the synthesis of biologically active compounds.

ACKNOWLEDGMENT: The Consiglio Nazionale delle Ricerche and the Ministero della Pubblica Istruzione (60%) are gratefully acknowledged.

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8. The utility of β -acetoxynitro precursors for the *in situ* generation of nitroalkenes has been previously experienced: D.H.R. Barton, S.Z. Zard, *J. Chem. Soc. Commun.*, 1985, 1098.
9. Satisfactory analytical and spectroscopic data have been obtained for all new compounds.
10. The ^1H NMR integration of both crude reaction mixtures reveals the presence of a minor product (20:1) identified as the *cis*-isomer.
11. The same ratio of cyclized products has been obtained by reaction of 1-phenyl-1-nitroethylene, generated *in situ* from 2-benzoyloxy-1-phenyl-1-nitroethane, and **1**. Therefore the possibility of stereochemical equilibration via a reverse Michael reaction in the cyclization step or through basic epimerization appears unlikely.